## Model Based Assessment of the Novel Use of Sour Water Stripper Vapor for NOx Control in CO Boilers

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# Abstract

CO boilers are commonly utilized in oil refineries to produce process steam by burning the low BTU, CO-containing gasses that are produced during regeneration of the fluid catalytic cracking (FCC) catalysts. Supplemental heating through fuel gas burners is necessary to carry out complete oxidation of the CO containing regeneration gases. The regeneration gases may contain varying concentrations of fuel nitrogen species including HCN, NH<sub>3</sub>, and NO, which can be sources of NO generation along with thermally produced NO in the CO boiler.

This paper discusses the results of a Computational Fluid Dynamics (CFD) based evaluation to guide the design of selective non-catalytic reduction (SNCR) for NOx control in two typical CO boiler configurations. In addition to the regeneration gases that are combusted in both boilers, a significant quantity of sour water stripper (SWS) vapor is injected into the combustion zone, subsequently generating fuel NOx emissions. The CFD simulations show the potential for reducing NOx emission by relocating a portion of the SWS vapor from the combustion zone to the post-combustion zone, where the SWS vapor acts as an SNCR reagent. Potential impacts on  $H_2S$  emissions by relocation of the SWS vapor are also examined.

# **1** Introduction

In the hydrocarbon processing industry, the operation of a fluid catalytic cracking unit (FCCU) produces low BTU, CO-containing gases. To reclaim the energy in these gases and reduce CO emission, a CO boiler is commonly included in the FCCU design to combust CO and to generate steam. Supplemental heating through fuel gas burners is necessary to carry out complete oxidation of the CO containing regeneration gases and to generate additional steam required by the refinery. Depending on the operating mode of the FCCU, the regeneration gases may contain varying concentrations of fuel nitrogen species including HCN, NH<sub>3</sub>, and NO, which can be sources of NO generation along with thermally produced NO in the CO boiler.

Sour water stripping is used to remove ammonia (NH<sub>3</sub>) and hydrogen sulfide (H<sub>2</sub>S) from sour water streams, which can come from distillation, fluid catalytic cracking, catalytic reforming, coker and acid gas removal units etc. in a refinery. Sour water stripper (SWS) vapor can be mixed with the regeneration gases, then combusted in a CO boiler. The presence of NH<sub>3</sub> in the SWS vapor has potential to generate a significant quantity of fuel NO<sub>x</sub> emissions in the CO boiler.

This paper discusses the results of a CFD-based evaluation to guide the design of selective noncatalytic reduction (SNCR) using SWS vapor as a SNCR reagent for NO<sub>x</sub> control in two typical CO boiler configurations. In both boilers, sour water stripper (SWS) vapor is mixed with the regeneration gas before the mixture is injected into the combustion zone to be combusted, subsequently generating a significant quantity of fuel NOx emissions. The CFD simulations evaluate the significant potential for NO<sub>x</sub> reduction by relocating a portion of the SWS vapor from the combustion zone to the post-combustion zone, where the SWS vapor acts as an SNCR reagent. Potential impacts on  $H_2S$  emissions by relocation of the SWS vapor are also exainned since the SWS vapor also contains  $H_2S$ .

# 2 Methods

### 2.1 CO Boiler Descriptions

Two typical refinery CO boiler configurations are shown in Figures 1 and 2. The first typical style, a "wall fired" CO boiler (shown in Figure 1), includes fuel gas burners that are fired horizontally from the end of the boiler, while regeneration gas mixed with SWS vapor is injected into the combustion zone from the top of the boiler. The CFD evaluation included the relocation of a portion of the SWS vapor from the regeneration gas to new SNCR ports located in the post combustion zone. Different SWS vapor rates to the combustion and SNCR zones were evaluated through the CFD simulations in order to determine optimum operation for NOx reduction.

A typical tangentially fired CO boiler configuration is shown in Figure 2. In this style of CO boiler, the regeneration gas ports and the fuel gas burners are located at the corners of the CO boiler, with the burner region of the boiler lined with refractory and the upper region of the boiler enclosed with bare waterwall tubes. The CFD evaluation for this style of CO boiler reviewed the relocation of a portion of the SWS vapor from the regeneration gas to new SNCR ports installed above the burner region where conditions are more favorable for SNCR chemistry. Different SWS vapor rates to the regeneration gas and SNCR ports were evaluated in order to optimize operation for NOx reduction.



Figure 1: A wall fired CO boiler



Figure 2: A tangentially fired CO boiler

#### 2.2 CFD Model Description

Reaction Engineering International (REI)'s proprietary CFD code, *ADAPT*, was used for the evaluation reported in this paper. *ADAPT* is a three dimensional, reacting CFD code that has been extensively utilized to simulate gas-fired combustion applications requiring an accurate treatment of chemistry and turbulence-chemistry interactions [1, 2].

The gas-phase fluid mechanics, heat transfer, thermal radiation and species transport are solved through conservation governing equations in an Eulerian framework. This software is a Reynolds Averaged Navier Stokes based CFD tool using a standard k- $\varepsilon$  two equation turbulence model. The code utilizes the discrete-ordinates model for radiation heat transfer [3, 4]. This method retains the directional dependency of the radiation intensity in a way that other flux models are unable to achieve, yet provides for a finite-difference or finite-volume solution that is more computationally efficient than zone methods and more deterministic than Monte Carlo methods.

The computational mesh is unstructured Cartesian with localized mesh refinement. Finite rate chemical kinetics are implemented through use of reduced mechanisms developed from the Computer Assisted Reduced Method (CARM) [5], based on steady state assumptions. The detailed mechanism used to develop the reduced mechanism is the GRI3.0 mechanism with addition of SNCR NO<sub>x</sub> chemistry. The reduced Mechanism used in this study contains 22 non-steady state species (H<sub>2</sub> H O<sub>2</sub> OH H<sub>2</sub>O HO<sub>2</sub> H<sub>2</sub>O<sub>2</sub> CH<sub>3</sub> CH<sub>4</sub> CO CO<sub>2</sub> CH<sub>2</sub>O CH<sub>3</sub>OH C<sub>2</sub>H<sub>2</sub> C<sub>2</sub>H<sub>4</sub> C<sub>2</sub>H<sub>6</sub> NH<sub>3</sub> NO N<sub>2</sub>O HCN HNCO N<sub>2</sub>). In-situ Adaptive Tabulation (ISAT) is used to reduce the computational time associated with integration of the finite rate chemical kinetics. Turbulence-chemistry interactions are represented using the eddy dissipation concept (EDC).

# **3** Results and Discussion

### 3.1 Wall Fired CO Boiler

The gas temperature profile under baseline operation is shown in Figure 3. Even though some temperature variations are observed, the overall gas temperatures between plane 1 and plane 3 are favorable for SNCR, with residence time between plane 2 and plane 3 adequate for SNCR chemistry. Figure 4 shows the CO concentration profile under baseline operation. Although the CO concentration is relatively high in the horizontal combustion region, the CFD predicted average CO concentration is very low after the gases make the 90° turn (i.e., plane 2). The predicted NO<sub>x</sub> concentration profile under baseline operation is shown in Figure 5. Most of the NO<sub>x</sub> is formed in the horizontal combustion section, with some formed between plane 1 and plane 2. The NO<sub>x</sub> chemistry is essentially quenched after plane 2.



Figure 3: Flue gas temperatures for baseline operation



Figure 4: Flue gas CO concentrations for baseline operation



Figure 5: Flue gas NO<sub>x</sub> concentrations for baseline operation

Figure 6 and Figure 7 show average flue gas temperatures and CO concentrations for simulations involving potential modifications to relocate a portion of SWS vapor to downstream SNCR ports. The influence of modifications on the gas temperature in the horizontal combustion section can be seen from Figure 6. However, downstream of plane 2, the average gas temperatures are similar for the modification cases and the baseline. Figure 7 also shows that the CO concentration is low for all the cases after plane 2. This suggests that the simulated modifications have negligible effects on the combustion efficiency. Figure 8 and Figure 9 show the NO<sub>x</sub> concentrations and NH<sub>3</sub> concentrations for the baseline in comparison with the modification cases. As can be seen from Figure 8, significant reduction in NO<sub>x</sub> emission is predicted for both cases involving relocation of SWS vapor to the SNCR ports. The NO<sub>x</sub> reductions come from both reduced NO<sub>x</sub> formation in the horizontal combustion zone due to less fuel-N species, and conversion of NO<sub>x</sub> to N<sub>2</sub> in the post combustion zone with SNCR chemistry using SWS vapor as an SNCR reagent. These results indicate that the relocation of SWS vapor to the post combustion SNCR zone is a very attractive approach to abate the NO<sub>x</sub> emission in this type of CO boiler. Since the SWS vapor also contains H<sub>2</sub>S, a potential concern is increased H<sub>2</sub>S emissions with this approach. This issue is discussed later in this paper



Figure 6: Average gas temperatures vs boiler length



Figure 7: Average CO concentrations vs boiler length



Figure 8: Average NO<sub>x</sub> concentrations vs boiler length



Figure 9: Average NH<sub>3</sub> concentrations vs boiler length

#### 3.2 Tangentially Fired CO Boiler

For the tangentially fired CO boiler, Figure 10 shows the baseline predicted gas temperatures, CO concentrations, and NO<sub>x</sub> concentrations before any proposed modifications. Based on predicted temperatures and CO concentrations, SNCR ports were simulated at a suitable location above the upper burner elevation. In this evaluation, a fixed portion of the total SWS vapor was mixed with the regeneration gas in the combustion zone, and incrementally increasing SWS vapor flow rates through the SNCR ports were simulated, from low (SNCR1) to median (SNCR2) and to high (SNCR 3). The predicted NO<sub>x</sub> and NH<sub>3</sub> concentrations as a function of boiler height are shown in Figure 12 and Figure 13. Significant NO<sub>x</sub> reduction is predicted at the different SWS rates to the SNCR ports. NO<sub>x</sub> reduction increases as the amount of SWS gas to the SNCR ports increases. However, NH<sub>3</sub> slip is also increased, primarily due to reduced flue gas temperature in this boiler. These results indicate that optimization of the quantity of SWS relocated to the SNCR ports will be dependent on what levels of ammonia slip are acceptable.



Figure 10: Gas temperatures, CO concentrations, and NO<sub>x</sub> concentration in a tangentially fired CO boiler under baseline operation



Figure 11: Average gas temperatures as a function of boiler height under baseline operation



Figure 12: Predicted NO<sub>x</sub> concentration as a function of boiler height



Figure 13: Predicted NH<sub>3</sub> concentration as a function of boiler height

#### 3.3 H<sub>2</sub>S Oxidation Chemistry

The SWS vapor may also contain H<sub>2</sub>S and the relocation of this vapor from the combustion zone to the SNCR ports in a CO boiler raises the potential concern of increased H<sub>2</sub>S emissions. A cursory evaluation was carried out to evaluate the potential for increased H<sub>2</sub>S emissions. Using a detailed mechanism based on GRI3.0, combined with the sulfur mechanism developed by Haynes [6], a series of isothermal plug flow calculations were carried out to evaluate the rate of oxidation of H<sub>2</sub>S over a relevant temperature range. The results, as shown in Figure 14, indicates a time lag associated with generation of the radical pool before the H<sub>2</sub>S oxidation takes off. The residence time needed at higher operating temperature for complete oxidation of the H<sub>2</sub>S is relatively short. The results show potential for significant H<sub>2</sub>S oxidation at typical SNCR zone temperatures for the residence times that exist in both CO boilers simulated here. However, quantitative predictions of H<sub>2</sub>S emissions should include the impacts of mixing, which can be achieved by integration of the chemical mechanism for H<sub>2</sub>S oxidation into the CFD model.



Figure 14: Isothermal plug flow calculation of H<sub>2</sub>S oxidation

## **4** Conclusions

Results of a CFD-based evaluation performed on two typical styles of CO boilers show that significant NOx reduction can be obtained by relocating SWS vapor from the combustion zone to downstream SNCR ports in the boiler. Potential NOx reduction is dependent on the flue gas temperature, available residence time, and the mixing of reagent and flue gas. NOx reduction was also impacted by the amount of SWS vapor that was directed to the SCNR ports, with higher NOx reduction seen with higher SWS rates to the SNCR ports. The overall NOx reduction may be limited, however, by NH<sub>3</sub> slip, as the modeling also showed increased NH<sub>3</sub> slip at higher SWS vapor rates to the SCNR ports, especially for boilers that have colder operating temperatures. Detailed chemical kinetics calculations to evaluate the impact of temperature on H<sub>2</sub>S oxidation indicate that there is significant H<sub>2</sub>S oxidation at typical SNCR zone temperatures. However, impacts of mixing also needs to be considered, ideally within a CFD simulation, to obtain accurate predictions of the impact of SWS relocation on H<sub>2</sub>S emissions.

## References

- Q. Tang, B. Adams, M. Bockelie, M. Cremer, M. Denison, C. Montgomery, A. Sarofim, D. Brown, "Towards Comprehensive CFD Modeling of Lean Premixed Ultra-Low NOx Burners in Process Heaters," AIChE 2005 Spring National Meeting (2005).
- [2] M. Cremer, D. Wang, M. Denison, B. Adams, "Demonstrated Implementation of Detailed Chemistry into CFD Simulations for Tailored Commercial Evaluation of Industrial Combustion Equipment," AFRC Industrial Combustion Symposium (2014).

- [3] Adams, B. R., and Smith, P. J., 1993, "Three-Dimensional Discrete-Ordinates Modeling of Radiative Transfer in a Geometrically Complex Furnace," Combustion Science and Technology, 88, pp.293-308.
- [4] Adams, B. R., and Smith, P. J., 1995, "Modeling Effects of Soot and Turbulence-Radiation Coupling on Radiative Transfer in Turbulent Gaseous Combustion," Combustion Science and Technology, 109, pp.121-140,.
- [5] J.-Y.Chen, A General Procedure for Constructing Reduced Reaction Mechanisms with Given Independent Relations. Combust. Sci. Technol. 57 (1988), 89–94.
- [6] Zhou, C., Sendt, K., and Haynes, B., "Experimental and kinetics modelling study of H<sub>2</sub>S oxidation" Proceedings of the Combustion Institute, 34 (2013), 625-632.

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